THE SYNTHESIS OF S-TRIAZOLO/1, 5-6/PYRIDAZINE 3-OXIDE DERIVATIVES

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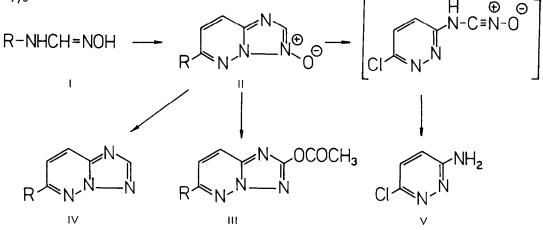
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In a recent paper the synthesis of 2-substituted s-triazolo/1, 5-a/pyridine 3-oxide and s-triazolo/1, 5-a/pyrimidine 3-oxide has been described 1 . To our knowledge, there is no general synthesis for the preparation of 2-unsubstituted s-triazoloazine 3-oxides with bridgehead nitrogen.

We wish to report the synthesis of 6-substituted s-triazolo/1, 5-b/pyridazine 3-oxide derivatives from 6-chloro-3-hydroxyiminomethyleneaminopyridazine. It represents a potential method for the preparation of other s-triazoloazine 3-oxides from heterocyclic hydroxyiminomethyleneamino derivatives. These were used as the starting material in a general synthesis of s-triazoloazines with the triazole ring fused to the azine ring through the N₂-C₃ bond².

The oxidation of I (R=6-chloropyridaziny1-3-) with bromine in acetic acid in the presence of sodium acetate results in the formation of 6-chloro-s-triazolo/1,5-b/pyridazine 3-oxide (II, R = CI, m.p. 218°, M^+ = 170, NMR spectrum in CDCl₃/TMS: T = 1.57 (s, H₂), 1.97 (d, H₈), 2.56 (d, H₇), J_{7.8} = 9.5 Hz) in 71 % yield.



The compound II is thermally stable. No deoxygenation was observed in boiling toluen **a** or xylene after 2 hours. In refluxing acetic anhydride it was transformed into 2-acetoxy-6-chloro-s-triazolo /1,5-b/pyridazine (III, R = Cl, m.p. 230°, $M^+ = 212$, decomp.) in 4 % yield. Nucleophilic substitution of chlorine at position 6 in 6-chloro-s-triazolo/1,5-b/pyridazine 3-oxide is taking place with hydrazine hydrate (98 %) in boiling ethanol giving 6-hydrazino-s-triazolo/1,5-b/pyridazine 3-oxide (II, R = NHNH₂ m.p. 230-232°, $M^+ = 166$, NMR spectrum in d_6 -DMSO/TMS: $\mathcal{T} = 1.78$ (s, H_2), 2.25 (d, H_8), 3.00 (d, H_7), $J_{7,8} = 9.5$ Hz) in 36 % yield. It was transformed with nitrous acid into 6-azido-s-triazolo /1,5-b/pyridazine 3-oxide (II, R = N₃, m.p. 130-134°, $M^+ = 177$, NMR spectrum in CDCl₃/TMS: $\mathcal{T} = 1.45$ (s, H_27 , 2.0 (d, H_8), 3.02 (d, H_7), $J_{7,8} = 9.5$ Hz) in 83 % yield, identical with the compound obtained from 6-chloro-s-triazolo/1,5-b/pyridazine 3-oxide and sodium azide in refluxing ethanol in 3,5 % yield.

Hydrogenation of II ($R = N_3$) in the presence of Pd/C (5 %) in methanol afforded 6-amino-s-triazolo/1, 5-b/pyridazine 3-oxide (II, $R = NH_2$, m.p. 259-263°) in 65 % yield, while in the presence of hydrochloric acid 6-amino-s-triazolo/1, 5-b/pyridazines was isolated, identical with an authentic sample prepared according to the lit.². Deoxygenation of II (R = CI or N_3) was achieved also with PCI₃ in chloroform giving s-triazolo/1, 5-b/pyridazine and 6-azido-s-triazolo/1, 5-b/pyridazine, respectively, identical with the compounds reported in the lit.².

Decomposition of triazole part of the molecule was observed when 6-chlaro-s-triazolo /1, 5-b/pyridazine 3-oxide (II, R = Cl) was refluxed either in 2 N hydrochlaric acid or in 2 N sodium hydroxide solution. 3-Amino-6-chlaropyridazine (V) was isolated as the only product. Sodium ethoxide or sodium thiophenolate afforded 3-amino-6-chlaropyridazine (V) together with a number of other unidentified products.

Satisfactory analytical data were obtained for all compounds.

References

1. T.L.Gilchrist, C.J.Harris, C.J. Moody, and C.W.Rees, J.C.S. Chem.Comm. <u>1974</u>, 486

2. S.Polanc, B.Verček, B.Šek, B.Stanovnik, and M.Tišler, J.Org.Chem., <u>39</u>, 2143 (1974).